9-(Triphenylphosphine-*P*)-arachno-6-thiadecaborane(11) Trichloromethane (1/1)

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Abstract

The title compound, $C_{18}H_{26}B_9PS.CHCl_3$ is shown to have an *arachno* ten-vertex {6-SB₉H₁₁} cluster structure with a PPh₃ ligand in the *exo*-9-position. The two bridging H atoms occupy asymmetric positions [average B—H_{bridge} values 1.21 (4) and 1.35 (4) Å] displaced away from the S atom.

Comment

Dicarbaborane chemistry dominates the molecular polyhedral chemistry of boron (Štíbr, 1992; Saxena & Hosmane, 1993). There is, however, tremendous potential for much added variety in other heteroborane chemistry, which is, by comparison, sparsely represented. The next best represented polyhedral heteroboranes are those of sulfur, and there is increasing interest in the investigation of this area (e.g. Jelínek et al., 1991). One current focus is on the ten-vertex arachno-6-thiadecaboranes, for both their intrinsic interest and as starting substrates for further syntheses (Ferguson et al., 1990; Holub et al., 1994; Adams et al., 1995). In this context, a structural report on [9-(Cy₂PPh)-arachno-6-SB₉H₁₁], (I) (where Cy is cyclohexyl) (Rosair et al., 1996), which alluded to an apparent uncertainty in the structure of its 9-(PPh₃) congener, (II) (Štíbr et al., 1996; Nestor et al., 1991), prompted us to report our structural work on compound (II).



The title compound (II), as isolated from the reaction of PPh₃ with *nido*-6-SB₉H₁₁ (Štíbr *et al.*, 1996), is seen to adopt an *exo*-(PPh₃) configuration, like the Cy₂PPh analogue, (I), and other 9-ligand derivatives that have been structurally characterized (Štíbr *et al.*, 1996; Hilty & Rudolph, 1979). The reported NMR parameters for

© 1998 International Union of Crystallography Printed in Great Britain – all rights reserved [9-(PPh₃)-arachno-6-SB₉H₁₁], as isolated from the thermolysis of [exo-9-{Cl-trans-(PPh₃)₂-cis-H₂Ir}-arachno-6-SB₉H₁₁] (Nestor *et al.*, 1991), for which the possibility of an endo-9-(PPh₃) placement was tentatively proposed (though not equivocally established), are somewhat different from those reported for the exo-9-(PPh₃) species structurally established here (Štíbr et al., 1996). The endo-9-(PPh₃) structure was initially proposed on the basis of a lower ¹H NMR shielding of the B9 proton (Nestor et al., 1991), but more recent systematic work (Štíbr et al., 1996) has shown that such a lower shielding is in fact quite general for the exo-structured compounds. On the other hand, the little information available (Štíbr et al., 1996) suggests that exo and endo isomers in these [9-L-arachno-6-EB9H11] systems (where E = NH as well as S) often do have very similar NMR properties.

The crystallographic data obtained in this study assist comparison with compounds with other ligands L in the *exo-9-L-arachno-6-SB*₉H₁₁ system (Table 2), of which $L = \text{NEt}_3$ (Hilty & Rudolph, 1979), (III), and MeCN (Štíbr *et al.*, 1996), (IV), have also been reported. There appear to be differential effects of the ligands on intracluster distances adjacent to the ligand site. Thus, distances from the substituted site B9 to its adjacent B atoms are shorter for L = MeCN, (IV), than for the stronger ligand NEt₃, (III), whereas those for the two reasonably strong phosphine ligands are intermediate between the two, though closer to the value for $L = \text{NEt}_3$, (III). Other interboron distances and S— B distances are very similar among all four compounds for equivalent sites.

An interesting general structural feature that now appears to be consistently emerging for all ten-vertex *arachno*-6-heterodecaboranes is a marked asymmetry of the bonding from the bridging H atoms to the B atoms that they bridge. These are the open-face positions α and β to the heteroatom. Thus, the results

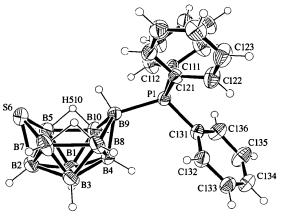


Fig. 1. Perspective view of a single molecule of (II) drawn with 40% probability displacement ellipsoids and with H atoms shown as small circles of arbritrary radii.

Acta Crystallographica Section C ISSN 0108-2701 © 1998 B1— B2—

B2—

B2— B3—

B3—

B3—

for the title compound (II) suggest longer B—H_{bridge} distances on the side adjacent (α) to the heteroatom [mean 1.35 (4) Å] compared to those remote (β) from the heteroatom [mean 1.21 (4) Å]. Similar differential effects are also observed in ten-vertex *arachno*-9,6-metallaheterodecaboranes (Kukina *et al.*, 1985; Jones *et al.*, 1994; Hilty, Thompson *et al.*, 1979; Faridoon *et al.*, 1989; Kim *et al.*, 1998).

Experimental

The title compound $[9-(PPh_3)-6-SB_9H_{11}]$, (II), was prepared according to the method of Štíbr *et al.* (1996) and recrystallized as colourless prisms from chloroform-hexane. Characterization data have been reported elsewhere (Štíbr *et al.*, 1996).

Crystal data

C ₁₈ H ₂₆ B ₉ PS.CHCl ₃	Cu $K\alpha$ radiation
$M_r = 522.08$	$\lambda = 1.54186 \text{ Å}$
Monoclinic	Cell parameters from 28
$P2_1/c$	reflections
a = 11.1590 (12) Å	$\theta = 35.3 - 39.7^{\circ}$
b = 25.112(2) Å	$\mu = 4.383 \text{ mm}^{-1}$
c = 9.6630(9) Å	T = 200 (2) K
$\beta = 90.641 (9)^{\circ}$	Prism
$V = 2707.7(5) \text{ Å}^3$	$0.41 \times 0.37 \times 0.31$ mm
Z = 4	Colourless
$D_x = 1.281 \text{ Mg m}^{-3}$	
D_m not measured	

Data collection

Stoe Stadi-4 four-circle	3131 reflections with
diffractometer	$I > 2\sigma(I)$
ω/θ scans	$R_{\rm int} = 0.022$
Absorption correction:	$\theta_{\rm max} = 64.75^{\circ}$
ψ scans (North, Phillips	$h = -13 \rightarrow 13$
& Mathews, 1968)	$k = -29 \rightarrow 0$
$T_{\rm min} = 0.199, \ T_{\rm max} = 0.251$	$l = -10 \rightarrow 11$
4341 measured reflections	2 standard reflections
4256 independent reflections	frequency: 60 min
	intensity decay: none

Refinement

 $(\Delta/\sigma)_{\rm max} = -0.021$ Refinement on F^2 $\Delta \rho_{\rm max} = 0.285 \ {\rm e} \ {\rm \AA}^{-3}$ $R[F^2 > 2\sigma(F^2)] = 0.055$ $wR(F^2) = 0.146$ $\Delta \rho_{\rm min} = -0.378 \ {\rm e} \ {\rm \AA}^{-3}$ S = 1.006Extinction correction: none 4256 reflections Scattering factors from 378 parameters International Tables for H atoms: see below Crystallography (Vol. C) $w = 1/[\sigma^2(F_o^2) + (0.058P)^2 +$ 3.8423*P*] where $P = (F_o^2 + 2F_c^2)/3$

Table 1. Selected bond lengths (Å)

B1—B2	1.742 (7)	B4—B10	1.740 (6)
B1—B5	1.769 (6)	B4—B8	1.744 (6)
B1—B4	1.771 (6)	B5B10	1.843 (6)
B1—B10	1.783 (6)	B5H510	1.33 (4)

-B3	1.792 (7)	S6—B2	1.913 (5)
-B3	1.741 (6)	S6—B5	1.913 (5)
-B7	1.882 (7)	S6—B7	1.932 (5)
-B5	1.886 (6)	B7—B8	1.856 (6)
-B7	1.764 (8)	B7—H78	1.36 (4)
-B4	1.773 (6)	B8—H78	1.21 (4)
-B8	1.781 (6)	B10H510	1.21 (4)

Table 2. Comparison of distances and angles involving the ligand-substituted B9 atoms in compounds $LB_9H_{11}S$

Distances	PPh ₃ , (II)	Cy ₂ PPh ^a , (I)	NEt ₃ , (III)	MeNC ^b , (IV)
B9—B4	1.719 (5)	1.732 (9)	1.723 (5)	1.742 (5)
B9—B8	1.861 (5)	1.896 (9)	1.904 (5)	1.872 (4)
B9—B10	1.888 (5)	1.897 (10)	1.904 (5)	1.872 (4)
B9—H9	1.12 (3)	1.10 (5)	1.09 (4)	1.05 (3)
Angles at B9 B4—B9—B8 B4—B9—B10 B8—B9—B10	58.1 (2) 57.5 (2) 104.4 (3)	57.2 (4) 57.2 (4) 102.5 (4)	56.9 (2) 57.0 (2) 103.4 (2)	57.8 (2) 57.8 (2) 104.4 (2)

Notes: (a) a common $B-H_{terminal}$ distance was refined. (b) There is a crystallographic mirror plane passing through atoms S6, B2, B4 and B9.

Phenyl rings were restrained to be flat and of overall $C_{2\nu}$ symmetry. All terminal and bridging borane H atoms were located from Fourier difference syntheses and all associated parameters were refined freely. Phenyl H atoms were included with a riding model in calculated positions [C—H = 0.95 Å and $U(H) = 1.2U_{eq}(C)$]. A disordered chloroform solvent molecule was modelled as two parts having site occupancies in the ratio 4:1 (occupancy parameters refined for both parts), and each part was treated with equivalent rigid bond and similarity restraints.

Data collection: *DIF*4 (Stoe & Cie, 1988*a*). Cell refinement: *DIF*4. Data reduction: *REDU*4 (Stoe & Cie, 1988*b*). Program(s) used to solve structure: *SHELXS*86 (Sheldrick, 1990). Program(s) used to refine structure: *SHELXL*93 (Sheldrick, 1993). Molecular graphics: *ORTEP*3 (Farrugia, 1997). Software used to prepare material for publication: *SHELXL*93.

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Supplementary data for this paper are available from the IUCr electronic archives (Reference: AB1503). Services for accessing these data are described at the back of the journal.

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3-Acetoxy-6-hydroxy-2,4-dimethoxyacetophenone[†]

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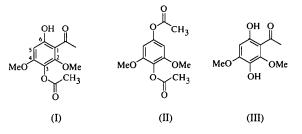
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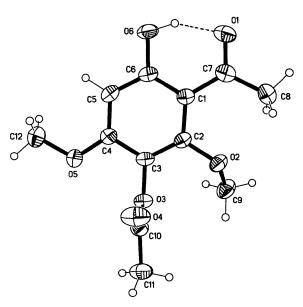
Abstract

The title compound, $C_{12}H_{14}O_6$, has been isolated from a Fries reaction on 3,6-dihydroxy-2,4-dimethoxyacetophenone. Its molecular structure contains an intramolecular hydrogen-bonded unit involving the -COCH₃ and -OH substituents. The best plane through the acetoxy group makes a dihedral angle of $88.74(5)^{\circ}$ with the plane of the aromatic ring.

Acetophenones are useful synthons for the preparation of a wide variety of polyphenolic compounds such as chalcones and flavones (Parmar et al., 1996). The synthesis of 3,6-dihydroxy-2,4-dimethoxyacetophenone, (III), was attempted via the Fries reaction on compound (II). However, this reaction failed to produce the desired product and instead the partially protected compound (I) (Fig. 1) was obtained as one of the side products; this latter compound has potential in the synthesis of bioactive flavonoids. The X-ray structure of (I) has been determined in order to obtain an unambiguous characterization.



The bond lengths and angles are unexceptional. The variation in C-O bond lengths within the methoxy groups is as expected, *i.e.* $O - C_{sp^3} > O - C_{sp^2}$ (e.g. Mukherjee et al., 1996). The methoxy substituent at C4 is almost coplanar with the aromatic ring, but that at C2 is twisted to give a C9-O2-C2-C3 torsion angle of $61.2(2)^{\circ}$. The acetoxy group is also twisted, making a dihedral angle of 88.7(1)° with the plane of the aromatic ring. The O1-C7-C1-C6-O6-H6 unit is essentially planar (r.m.s. deviation 0.018 Å) as a result of



† Alternative name: 3-acetyl-4-hydroxy-2,6-dimethoxyphenyl ethano- Fig. 1. View of the title molecule showing the atomic numbering. Displacement ellipsoids are drawn at the 50% probability level.

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